

Unexpected Orthogonality of S-Benzoxazolyl and S-Thiazolynyl Glycosides: Application to Expedient Oligosaccharide Assembly

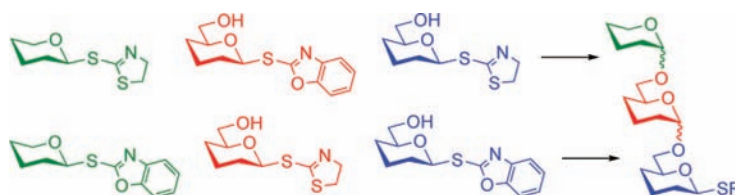
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ABSTRACT



Thorough mechanistic studies of the alkylation pathway for the activation of glycosyl thioimidates have led to the development of the “thioimide-only orthogonal strategy”. Discrimination among S-thiazolynyl (STaz) and S-benzoxazolyl (SBox) anomeric leaving groups was achieved by fine-tuning of the activation conditions. Preferential glycosidation of a certain thioimide is not simply determined by the strength of activating reagents; instead, the type of activation—direct vs indirect—comes to the fore and plays the key role.

Traditional linear approaches to oligosaccharide assembly are often cumbersome, and consequently, the availability of complex glycostructures remains insufficient to address the challenges of modern glycosciences.¹ Recent improvements in strategies for oligosaccharide assembly, have significantly shortened the number of synthetic steps required by minimizing protecting group manipulations between glycosylation steps.² One of the most flexible assembly strategies is the orthogonal concept.³ Unlike the armed–disarmed approach,⁴ the orthogonal activation is not reliant on the nature of the protecting groups,

which can interfere with stereoselectivity. The only requirement for the orthogonal approach is a set of two orthogonal leaving groups and a pair of suitable activators. Unfortunately, this simple concept is still limited to the following two examples: Ogawa’s S-ethyl and fluoride,³ and thioimide S-thiazolynyl (STaz) and S-alkyl/aryl.⁵ In addition, a related, albeit less flexible, semiorthogonal approach with the use of S-ethyl and O-pentenyl glycosides was reported⁶ and was recently extended to fluoride/pentenyl leaving groups.⁷ Overall, the orthogonal

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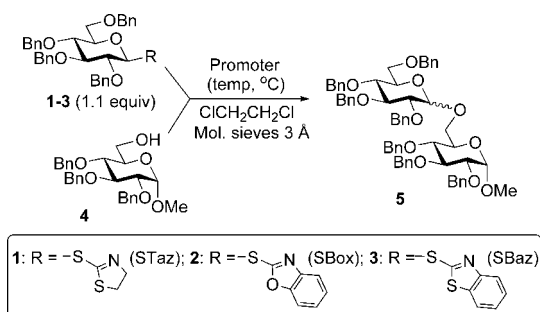
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strategy is an excellent concept for flexible sequencing of oligosaccharides that remains underexplored, with too few examples to become universal.

The excellent glycosyl donor properties of glycosyl thioimides and their unique activation conditions have led to a number of useful developments for oligosaccharide synthesis, including the orthogonal approach.⁵ From our previous studies, we had determined the *S*-benzoxazolyl (SBox) glycosyl donors⁸ to be significantly more reactive than their STaz counterparts,⁵ and although direct selective activations of the SBox donors over STaz acceptors were reported in the presence of Cu(OTf)₂,⁵ no comprehensive side-by-side comparisons had been performed.

With the main purpose of determining relative reactivity patterns of thioimides, we set up a series of glycosidations including STaz,⁵ SBox,⁸ and structurally related Mukaiyama's *S*-benzothiazolyl (SBaz) derivatives.⁹ All reactions of benzylated glycosyl donors **1–3** with glycosyl acceptor **4** promoted with MeOTf were very effective, and disaccharide **5** was obtained in high yields (Table 1). Although the

Table 1. Alkylation-Initiated Glycosidation of Benzylated Thioimides **1–3**



entry	donor	promoter (equiv, temp)	time (h)	product 5 (yield (%), α/β)
1	1	MeOTf (3, rt)	0.75	87, 1.4/1
2	2	MeOTf (3, rt)	0.33	88, 1.6/1
3	3	MeOTf (3, rt)	0.58	89, 1.6/1
4	1	MeI (9, rt)	120	89, 8.0/1
5	2 or 3	MeI (9–15, rt)	120	no reaction
6	1	BnBr (3, 55 °C)	24	90, 3.5/1
7	2	BnBr (3–9, 55 °C)	120	traces
8	3	BnBr (3–9, 55 °C)	120	no reaction

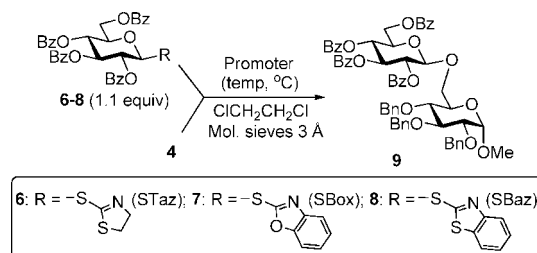
disparity among the reaction times was unremarkable, it was evident that glycosidation of the STaz glycoside **1** (entry 1) was about two times slower than that of its SBox counterpart **2** (entry 2), with the reactivity of the SBaz glycoside **3** (entry 3) somewhere in the middle.

To gain better control of the glycosylations and to achieve a more precise differentiation of reactivity between the different thioimides, we turned to investigating other activators. Among

these, results obtained with common alkylating (and acylating) reagents were of particular attractiveness. For example, MeI was only effective for glycosidation perbenzylated STaz glycoside **1** (entry 4). Surprisingly, when these reaction conditions were applied to the glycosidation of expectedly more reactive SBox and SBaz glycosides, no glycosylation took place (entry 5). Also, BnBr was only effective for glycosidation STaz derivative **1**, whereas glycosyl donors **2** or **3** gave only trace amounts of products (entries 6–8).

Subsequently, similar observations have been made with less reactive perbenzoylated glycosyl donors. Thus, all reactions of benzoylated glycosyl donors **6–8** with glycosyl acceptor **4** promoted with MeOTf were very effective, and disaccharide **9** was obtained in high yields (Table 2, entries

Table 2. Alkylation-Initiated Glycosidation of Benzoylated Thioimides **6–8**



entry	donor	promoter (equiv, T (°C))	time (h)	product, 9 (yield %)
1	6	MeOTf (3, rt)	2	97
2	7	MeOTf (3, rt)	1	95
3	8	MeOTf (3, rt)	2	84
4	6–8	MeI (9–15, rt)	120	no reaction
5	6	BnBr (3, 55)	72	79
6	7 or 8	BnBr (3–9, 55)	120	no reaction

1–3). With MeI though, no glycosidation of benzoylated glycosyl donors **6–8** took place (entry 4). BnBr was only effective for the STaz derivative **6**, (entry 5), whereas glycosidation of glycosyl donors **7** or **8** gave no products (entry 6). It should be noted that all reactions with benzylated glycosyl donors (Table 1) were significantly faster than those with their benzoylated counterparts (Table 2).

This discovery signified the gap in our understanding of the thioimide activation and created a basis for the development of the STaz-SBox orthogonal strategy. The uniqueness of this approach would be that both leaving groups employed are of essentially the same class. Although certain pathways for the activation of thioimides were postulated (Scheme 1),^{10,11} little proof had been available until our recent mechanistic study,⁸ wherein we showed that MeOTf-promoted activation of the SBox glycosyl donor **10** proceeds via the anomeric sulfur atom (direct activation). This was confirmed by isolating the departed *S*-methylated aglycon MeSBox (**11**, Scheme 1).

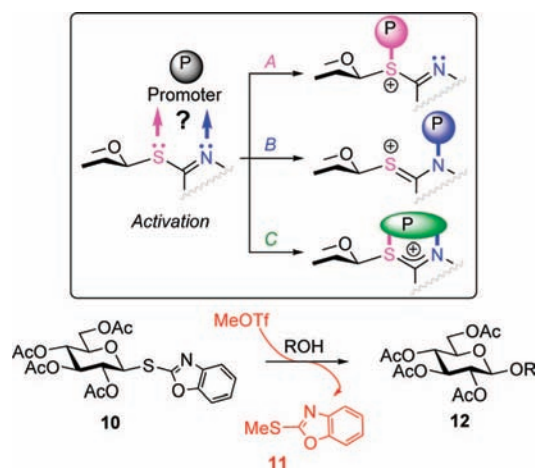
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Scheme 1. Activation of Thioimides: A Study with SBox Glycoside **10**⁸



Since the exact nature of the STaz activation was not known, our working hypothesis for this study was based on our previous findings with the SBox moiety, taking into consideration the structural differences between the two moieties. We anticipated that the activation of the STaz moiety proceeds via the nitrogen (remote activation), as opposed to the direct activation of the SBox moiety. This remote activation of STaz would lead to a marginally slower reaction with a powerful promoter, such as MeOTf (refer to Tables 1 and 2). When weak alkylating reagents are used (MeI, BnBr), a powerful nucleophile is needed to replace the iodine or bromine, respectively. Evidently, this can be predominantly achieved with STaz glycosides that bear the reactive nitrogen atom, but not with the SBox glycosides, which can only be activated via the exocyclic sulfur⁸ (Figure 1). For comparison, previous reports suggest that SPh

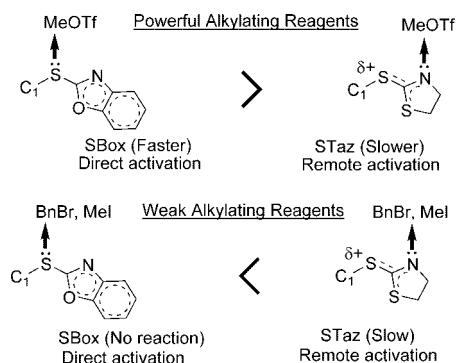
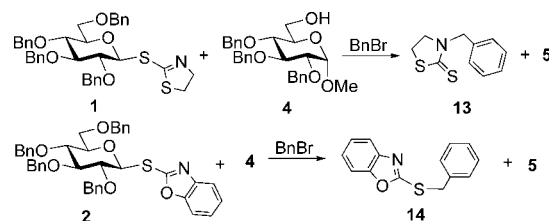


Figure 1. Working hypothesis for thioimide activation.

glycosides do not react with MeI (direct activation), while S-pyridyl derivatives do (remote activation is postulated).¹¹

The credibility of this working hypothesis was verified by a series of experiments in which glycosyl donor **1** was reacted with the standard acceptor **4** in the presence of BnBr (Scheme 2). Upon disappearance of **1**, the reaction mixture

Scheme 2. Mechanism of Thioimide Activation with BnBr



was concentrated in vacuo, and the high running UV-active spot, corresponding to the benzylated aglycon, was separated from the disaccharide product **5** by column chromatography on silica gel. The structure of the isolated alkylated aglycone was assigned as thioamide **13** (BnNTaz) by X-ray (see the Supporting Information) and UV ($\lambda = 277$ nm, $C=S$).

A similar reaction between **2** and **4** was significantly slower; nevertheless, we succeeded in isolating the departed aglycon, which was assigned as thioimide **14** (BnSBox) by X-ray (see the Supporting Information) and UV ($\lambda = 280, 287$ nm, $C=N$), as anticipated. To exclude the impact of tautomerization of the products both reactions were monitored by HPLC (see the Supporting Information).

With a better understanding of the mechanistic pathway, we were well positioned to undertake further studies of expeditious oligosaccharide assembly. First, the activation of STaz donors **1** and **6** over SBaz/SBox acceptors **15a–c** was investigated in the presence of BnBr or MeI. As a result, disaccharides **16a–c** were obtained in good yields (70–82%, Table 3). Conversely, SBox donor **7** was also activated over the STaz acceptor **15d**. This activation could be accomplished in the presence of a variety of activators, among which $Bi(OTf)_3$ was the most promising; disaccharide **16d** was obtained in 69% yield. The terminal SBox moieties of **16b** or **16c** could be glycosidated either with model acceptor **4** or with STaz acceptor (**15d**). Conversely, the STaz moiety of **16d** could be glycosidated with **4** or activated over the SBox acceptor (**15c**). The resulting trisaccharides **17a–d** were isolated in 62–92% yields (Table 3), with 1,6-anhydro and hemiacetal being the major byproduct identified. Among trisaccharides generated, **17c** and **17d** can be used in subsequent sequencing.

In conclusion, a mechanistic study of the alkylation pathway for the activation of glycosyl thioimides has led to the development of the “thioimide-only orthogonal strategy”. The synthesis of trisaccharides **17a–d** clearly illustrates the entirely orthogonal character of the SBox and STaz derivatives. Further studies and application of the new concept to glycosylation of secondary glycosyl acceptors are currently being pursued in our laboratory.

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Table 3. Orthogonal Activations for Oligosaccharide Synthesis

entry	donor	acceptor A	promoter A ^a	disaccharide (yield, α/β)	acceptor B	promoter B ^a	trisaccharide (yield, α/β)
1			BnBr	 (70%, β only)		AgBF ₄	 (75%, β only)
2			MeI	 (82%, 6/1)		AgOTf	 (92%, 1.6/1)
3			BnBr	 (76%, β only)		AgOTf	 (72%, β only)
4				 (76%, β only)		Bi(OTf) ₃	 (62%, β only)
5			Bi(OTf) ₃	 (69%, β only)		AgBF ₄	 (81%, β only)
6				 (69%, β only)		BnBr	 (67%, β only)

^a Conditions: All glycosylations were performed in 1,2-dichloroethane in the presence of molecular sieves (3 Å). Promoters: BnBr (3 equiv, 55 °C); AgBF₄ (3 equiv, rt); MeI (9 equiv, rt); AgOTf (2 equiv, rt); Bi(OTf)₃ (3 equiv, 0 °C → rt). Reaction time: **16a**, 24 h; **16b**, 16 h; **16c**, 24 h; **16d**, 1 h; **17a**, 30, 30, and 40 min (entries 1, 3, and 5, respectively); **17b**, 15 min; **17c**, 2 h; **17d**, 36 h.

Supporting Information Available: Experimental procedures, ¹H and ¹³C NMR spectra for all new compounds, and the crystal structure and geometrical parameters for **13**

and **14**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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